

FORM PTO-1350 (Modified)  
(REV 11-98)

U.S. DEPARTMENT OF COMMERCE PATENT AND TRADEMARK OFFICE

ATTORNEY'S DOCKET NUMBER

TRANSMITTAL LETTER TO THE UNITED STATES  
DESIGNATED/ELECTED OFFICE (DO/EO/US)  
CONCERNING A FILING UNDER 35 U.S.C. 371

WLJ.059

U.S. APPLICATION NO. (IF KNOWN, SEE 37 CFR

09/ 674 659

INTERNATIONAL APPLICATION NO.

PCT/GB00/00793

INTERNATIONAL FILING DATE

6 MARCH 2000

PRIORITY DATE CLAIMED

4 MARCH 1999

TITLE OF INVENTION

GAS GENERATION SYSTEM

APPLICANT(S) FOR DO/EO/US

Jyoti Kiron BHARDWAJ



Applicant herewith submits to the United States Designated/Elected Office (DO/EO/US) the following items and other information:

1. ☒ This is a **FIRST** submission of items concerning a filing under 35 U.S.C. 371.
2. ☐ This is a **SECOND** or **SUBSEQUENT** submission of items concerning a filing under 35 U.S.C. 371.
3. ☐ This is an express request to begin national examination procedures (35 U.S.C. 371(f)) at any time rather than delay examination until the expiration of the applicable time limit set in 35 U.S.C. 371(b) and PCT Articles 22 and 39(1).
4. ☒ A proper Demand for International Preliminary Examination was made by the 19th month from the earliest claimed priority date
5. ☒ A copy of the International Application as filed (35 U.S.C. 371 (c) (2))
  - a. ☒ is transmitted herewith (required only if not transmitted by the International Bureau).
  - b. ☐ has been transmitted by the International Bureau.
  - c. ☐ is not required, as the application was filed in the United States Receiving Office (RO/US).
6. ☐ A translation of the International Application into English (35 U.S.C. 371(c)(2)).
7. ☒ A copy of the International Search Report (PCT/ISA/210).
8. ☐ Amendments to the claims of the International Application under PCT Article 19 (35 U.S.C. 371 (c)(3))
  - a. ☐ are transmitted herewith (required only if not transmitted by the International Bureau).
  - b. ☐ have been transmitted by the International Bureau.
  - c. ☐ have not been made; however, the time limit for making such amendments has NOT expired.
  - d. ☐ have not been made and will not be made.
9. ☐ A translation of the amendments to the claims under PCT Article 19 (35 U.S.C. 371(c)(3)).
10. ☐ An oath or declaration of the inventor(s) (35 U.S.C. 371 (c)(4)).
11. ☐ A copy of the International Preliminary Examination Report (PCT/IPEA/409).
12. ☐ A translation of the annexes to the International Preliminary Examination Report under PCT Article 36 (35 U.S.C. 371 (c)(5)).

## Items 13 to 20 below concern document(s) or information included:

13. ☐ An Information Disclosure Statement under 37 CFR 1.97 and 1.98.
14. ☐ An assignment document for recording. A separate cover sheet in compliance with 37 CFR 3.28 and 3.31 is included.
15. ☒ A **FIRST** preliminary amendment.
16. ☐ A **SECOND** or **SUBSEQUENT** preliminary amendment.
17. ☐ A substitute specification.
18. ☐ A change of power of attorney and/or address letter.
19. ☐ Certificate of Mailing by Express Mail
20. ☐ Other items or information:

U.S. APPLICATION NO. (IF KNOWN, SEE 37 CFR <div style="font-size: 24pt; font-weight: bold; text-align: center;">097 674659</div>	INTERNATIONAL APPLICATION NO. <div style="font-weight: bold;">PCT/GB00/00793</div>	ATTORNEY'S DOCKET NUMBER <div style="font-weight: bold;">WLJ-059</div>		
The following fees are submitted: <b>BASIC NATIONAL FEE ( 37 CFR 1.492 (a) (1) - (5) ) :</b> <input type="checkbox"/> Neither international preliminary examination fee (37 CFR 1.482) nor international search fee (37 CFR 1.445(a)(2)) paid to USPTO and International Search Report not prepared by the EPO or JPO ..... <b>\$1,000.00</b> <input checked="" type="checkbox"/> International preliminary examination fee (37 CFR 1.482) not paid to USPTO but International Search Report prepared by the EPO or JPO ..... <b>\$860.00</b> <input type="checkbox"/> International preliminary examination fee (37 CFR 1.482) not paid to USPTO but international search fee (37 CFR 1.445(a)(2)) paid to USPTO ..... <b>\$710.00</b> <input type="checkbox"/> International preliminary examination fee paid to USPTO (37 CFR 1.482) but all claims did not satisfy provisions of PCT Article 33(1)-(4) ..... <b>\$690.00</b> <input type="checkbox"/> International preliminary examination fee paid to USPTO (37 CFR 1.482) and all claims satisfied provisions of PCT Article 33(1)-(4) ..... <b>\$100.00</b>		<b>CALCULATIONS</b> PTO USE ONLY		
<b>ENTER APPROPRIATE BASIC FEE AMOUNT =</b>		<b>\$860.00</b>		
Surcharge of <b>\$130.00</b> for furnishing the oath or declaration later than _____ months from the earliest claimed priority date (37 CFR 1.492 (e)). <input type="checkbox"/> 20 <input type="checkbox"/> 30		<b>\$0.00</b>		
CLAIMS	NUMBER FILED	NUMBER EXTRA	RATE	
Total claims	6 - 20 =	0	x \$18.00	
Independent claims	1 - 3 =	0	x \$80.00	
Multiple Dependent Claims (check if applicable).			<input type="checkbox"/> \$0.00	
<b>TOTAL OF ABOVE CALCULATIONS</b>			<b>\$860.00</b>	
Reduction of 1/2 for filing by small entity, if applicable. Verified Small Entity Statement must also be filed (Note 37 CFR 1.9, 1.27, 1.28) (check if applicable).			<input type="checkbox"/> \$0.00	
<b>SUBTOTAL</b>			<b>\$860.00</b>	
Processing fee of <b>\$130.00</b> for furnishing the English translation later than _____ months from the earliest claimed priority date (37 CFR 1.492 (f)). <input type="checkbox"/> 20 <input type="checkbox"/> 30			<b>\$0.00</b>	
<b>TOTAL NATIONAL FEE</b>			<b>\$860.00</b>	
Fee for recording the enclosed assignment (37 CFR 1.21(h)). The assignment must be accompanied by an appropriate cover sheet (37 CFR 3.28, 3.31) (check if applicable).			<input type="checkbox"/> \$0.00	
<b>TOTAL FEES ENCLOSED</b>			<b>\$860.00</b>	
			Amount to be: refunded	\$
			charged	\$
<input checked="" type="checkbox"/> A check in the amount of <b>\$860.00</b> to cover the above fees is enclosed.				
<input type="checkbox"/> Please charge my Deposit Account No. _____ in the amount of _____ to cover the above fees. A duplicate copy of this sheet is enclosed.				
<input checked="" type="checkbox"/> The Commissioner is hereby authorized to charge any fees which may be required, or credit any overpayment to Deposit Account No. <b>50-0238</b> A duplicate copy of this sheet is enclosed.				
<b>NOTE:</b> Where an appropriate time limit under 37 CFR 1.494 or 1.495 has not been met, a petition to revive (37 CFR 1.137(a) or (b)) must be filed and granted to restore the application to pending status.				
SEND ALL CORRESPONDENCE TO:				
ADAM C. VOLENTINE JONES VOLENTINE, LLC 12200 SUNRISE VALLEY DRIVE, SUITE 150 RESTON, VA 20191  TEL. NO. (703) 715-0870 FAX NO. (703) 715-0877			 SIGNATURE  ADAM C. VOLENTINE NAME  33289 REGISTRATION NUMBER  NOVEMBER 3, 2000 DATE	

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

In Re Patent Application of :  
Jyoti Kiron BHARDWAJ : Attn: Applications Branch  
Serial No. [NEW] : Attorney Docket No.: WLJ.059  
Filed: November 3, 2000 :  
Title: GAS GENERATION SYSTEM

**PRELIMINARY AMENDMENT**

Honorable Commissioner of  
Patents and Trademarks,  
Washington, D.C. 20231

Sir:

Preliminary to the examination of the above-identified application, please enter the following amendments and consider the following remarks:

**IN THE CLAIMS:**

Please cancel claims 7-9 without prejudice.

Please amend the claims as follows:

Claim 3, line 1, delete "or claim";

line 2, delete "2".

Claim 4, line 1, delete "or claim";

line 2, delete "2".

Claim 5, line 1, delete "or claim";

line 2, delete "2".


Claim 6, line 3, change "any one of claims 1 to 5" to --claim 1--.

**REMARKS**

By this Preliminary Amendment, claims 7-9 have been canceled, and claims 3, 4, 5, and 6 have been amended to eliminate the multiple dependent claims. Entry of this Preliminary Amendment is respectfully requested.

Respectfully submitted,

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Date: November 3, 2000

"Gas Generation System"

Industrial demands for alternative process gases for dry (plasma and non-plasma) process systems has led to the investigation of a number of different chemistries.

- 5 Several different highly-reactive halogen-based chemistries are known to be likely candidates to achieve an improved process capability including etch rate, selectivity etc. Gases in this category include pure-halide (inter-halogenated and molecular halogen) compounds and suffer
- 10 from increased cost (of the delivery and abatement system as well as precursor itself), greater health and safety risks, transportation difficulty (and cost) as well as relatively poor commercial availability.

- According to the invention there is provided a gas
- 15 generator system comprising supply sources for halogenated gases, including pure molecular halogens, connected into a gas reaction chamber or chamber system enabling generation of a predetermined gas for localised use in a subsequent process.

- 20 In the preferred arrangement the reaction chamber has a valved outlet for direct supply of the generated gas to a single or multiple chamber processing tool or process chamber.

- "Localised" (or point of use) means that the delivery
- 25 system is located near to a process chamber or a number of chambers or number of systems near to one another, so that the gases created can be delivered directly to the chamber

or system for immediate use rather than being created off-site and transported in a suitable container for subsequent introduction into the apparatus.

More than one generator may be used for each process  
5 chamber.

In one embodiment, this invention will provide for the localised generation of reactive process gases. These can be generated local to each process tool through the direct combination of different precursor gases under controlled  
10 temperature and pressure reaction conditions. Various inter-halogen and compound halogenated gases can be generated in this way. This system can include a feedback system to precisely control the precursor that is being generated.

15 A second embodiment of this invention will provide for the local on-board synthesis of reactive precursors for plasma and non-plasma processing which include precursors which would otherwise not be readily available due to instability or short-lifetime or difficulty in commercial  
20 manufacture or supply. This allows for the direct synthesis and reactive investigation of new groups of halide precursors for semiconductor or non-semiconductor processing.

A third embodiment of this invention provides for the synthesis of precursors that replace directly or indirectly  
25 (by simulating the behaviour) the gases that are either in use or have been used in industry and are subject to restriction in use or withdrawal by new environmental

legislation. The Montreal Protocol defines environmentally unfriendly chemicals including those that have already been restricted or those that will be restricted in application in the near future. This will impact on the availability of the gases as production is restricted and also on the increasing necessity for abatement during even limited use. Indeed, much recent work has focused on the search for suitable alternative replacement gases (Fracassi and d'Agostino, J.Vac.Sci.Technol.B 16(4), Jul/Aug. 1998). With suitable abatement means, the synthesis of the material on demand may well be an acceptable means of continuing to use existing chemistries. Commercially this may be the only one of a few solutions available.

The invention further extends to a method of generating gases for localised use in a subsequent process by utilising the system of this invention as defined hereinbefore and using the gas so generated directly in the subsequent process.

The direct combination gas generator can produce a gas in the process chamber by reacting suitable precursor gases possibly with other materials that may be preloaded within a temperature and pressure controlled reaction chamber that is local to a processing tool or chamber. The precursor gases may not react with the substrate being processed, or indeed be desired in the processing tool individually. The design of the system avoids the occurrence in the processing tool or chamber of any possible adverse

reactions, during the combination of the precursor gases, that might otherwise prejudice the overall process. Bypass of the process chamber to an abatement tool inhibits the pre-reaction gas mixture from adversely influencing the process in the process chamber. The precursor gases are redirected into the process chamber by the process controller after the required flow and composition has stabilised or wafer loading is complete. The bypass function may also be used during wafer transfer to maintain constant supply quality and conditions.

The precursors for the reaction chamber include any of the following gases:-

1. at least one halogen selected from  $F_2$ ,  $Cl_2$ ,  $Br_2$ ,  $I_2$  as the required halogen source;
2. additional gases from  $N_2$ ,  $H_2$ ,  $O_2$ ;
3. additional gases from He, Ar, Kr, Xe.

The generator may further comprise any one or more of the following components:-

1. electrode materials selected from C, B, S, Si, Ge, P in either elemental or appropriate compound form. Where appropriate, the electrode material may be porous to increase the surface reaction area. Metal-containing electrodes may also be used. The electrodes may be independently temperature controlled and electrically and/or magnetically biased;
2. appropriate catalysts (product generation depend-



ent);

3. gas analysers such as infra-red spectroscopes and/or mass spectrometers;
4. a feedback control system which senses the output gas and adjusts reaction parameters (gas flow rates, temperatures, pressure etc) to achieve the desired output concentration;
5. an appropriate reaction vessel or chamber. A number of chambers may be used in series and/or parallel to generate precursors requiring more complex generation mechanisms;
6. an appropriate vacuum control system, valving and gauging, including means for bypassing and purging;
7. an appropriate abatement system.

The generator can operate within the following temperature and pressure range: room temperature to several hundred degrees C at several Torr to atmospheric pressure. In particular the reaction chamber may be operated at or near atmospheric pressure, going up the range from several Torr to 760 Torr. The reaction chamber temperature can be controlled at between ambient room temperature up to 800°C generally, but probably will lie within the range of 100 - 500°C. Differing temperatures may be maintained in at least 2 separate zones of the reaction chamber system. For certain processes, elements of the system may be below room temperature for product purity control. The temperature

control of any electrode may be separate from the vessel and the electrode may be independently biased either electrically and/or magnetically. This or other biasing means may be used to generate a plasma.

5       The supply to the gas reaction generator of high purity gases removes the need to polish the generated gas to remove unwanted impurities before passing into the process chamber. Mass flow controllers may be used to precisely meter the flow of the supply gases.

10       The valved control of the flow into the process chamber (if necessary augmented by a control system) ensures that the generator does not experience the low pressure (vacuum) at the process chamber. The preferred method of production of fluorine is by fluoride molten  
15 electrolytic generators which can be installed locally to the tool and overcomes the need to obtain high purity 100% fluorine in cylinders. Other means of  $F_2$  generation can be appropriately used. The preferred method of supply of  $Cl_2$  is from either an electrolytic generator or from high-pressure  
20 cylinders, both of which are readily available. Similarly, appropriate sources of  $Br_2$ ,  $I_2$  may be used.

Multiple or singular gas inlets and/or either two or more electrodes (or a single composite electrode) as necessary may be used to generate some of the more complex  
25 halide species. Electrodes within the reaction chamber may be electrically and/or magnetically biased.

The reaction chamber can be formed from high purity

materials (such as those sold under the Trade Marks Monel (nickel/copper/iron alloy), Inconel (nickel/chromium/iron alloy) and Hastalloy (nickel/molybdenum/chromium/manganese/iron alloy)) which would not be financially feasible with large scale generation systems

The concept of local generation allows the production of the desired process gases to a high purity, although the specific reaction products may be a family of compounds not just one specific product. This can be achieved at a reasonable cost and low risk. The addition of this gas generator, feeding a process tool, allows a novel capability to introduce different generated process gases into the process chamber where, for many different reasons, the option of using the gas directly would not be possible.

The gas generator is able to provide a local supply of various gases on demand, such as :-

1. Inter-halogen compounds including  $\text{Cl}_x\text{F}_y$ ,  $\text{F}_x\text{Br}_y$ ,  $\text{Cl}_x\text{Br}_y$  and  $\text{F}_x\text{Cl}_y\text{Br}_z$  (where  $x$ ,  $y$  and  $z$  are integers). An  $\text{I}_2$  source could also be added to provide additional I-containing compound halogens;
2. Groups of halide precursors for plasma and non-plasma processing which include precursors which would otherwise be unavailable due to instability or short-lifetime or difficulty in commercial manufacture or supply;
3. Freons or halo-carbides including  $\text{C}_x\text{F}_y$ ,  $\text{C}_w\text{F}_x\text{Cl}_y$ ,  $\text{C}_w\text{F}_x\text{Br}_y$ ,  $\text{C}_w\text{F}_x\text{Cl}_y\text{Br}_z$ ,  $\text{C}_w\text{Cl}_x$ ,  $\text{C}_w\text{Cl}_x\text{Br}_y$ ,  $\text{C}_w\text{F}_x\text{Br}_y$ ,  $\text{C}_w\text{Br}_y$

(where  $w$ ,  $x$ ,  $y$  and  $z$  are integers). An  $I_2$  source could also be added to provide additional I-containing halo-carbides;

4. Other reactive halides where the C is replaced by N or B;
5. Other reactive halides where any C is replaced by combination of C with N, B and/or H;
6. Metal halides, where the C source is replaced by an appropriate metal-containing electrode, (including W, Mo, Ti and Ta);
7. Halide complexes of any combination of the groups listed in 3,4,5 and 6 above.

Ideal gases for generation include  $ClF$ ,  $ClF_3$ ,  $ClF_5$ ,  $BrF_3$ ,  $WF_6$ ,  $MoF_6$ ,  $TiF_4$ ,  $TiCl_4$ ,  $TaF_5$ ,  $BrF$ ,  $CF_4$ ,  $C_2F_6$ ,  $C_3F_8$ ,  $CHF_3$ ,  $CF_2H_2$ ,  $CCl_2F_2$ ,  $CCl_4$ ,  $CF_3Br$  and  $NF_3$ , and related compounds. An appropriate selection of precursors can also be used to synthesise or generate organo-metallic compounds

The generator of the invention offers many advantages over direct gas delivery. For some of the gases the reaction gas generator can operate at or near atmospheric pressure, thereby eliminating the need for high-pressure regulators on the system. Installation and storage of hazardous gases is reduced. The local generation, on demand, of the hazardous gases eliminates long (and usually high-pressure) gas lines from a central store to the installation and minimises associated safety risks. There would be a significant reduction in the installation costs

due to the reduced level of plumbing of additional gases and the associated safety requirements. The maintenance of the complete system is eased by the absence of any of the generated gases when the system is not operating. The  
5 quantity of the generated gases can be regulated to that required for the specific application so that the gas consumption is optimised and excess generated gas avoided. Some gases may require higher generation pressures (above atmospheric) and hence necessitate the use of pressure  
10 regulatory devices.

The invention may be performed in various ways and preferred embodiments thereof will now be described, by way of example, with reference to the accompanying drawings, in which:-

15 Figure 1 is a diagrammatic illustration of a general gas generation system of this invention;

Figure 2 indicates various input and control features of the system of Figure 1;

Figure 3 shows a modified system for that indicated in  
20 Figure 2; and

Figures 4, 5 and 6 illustrate still further modifications of the general gas generation system illustrated in Figure 1.

In the arrangement shown in Figure 1 precursor gases  
25 are passed from suitable supply sources 1 and 2 to a reaction chamber 3 where the gases are combined under controlled conditions. The reaction product is then fed to

a process chamber 4 where a dry process utilising that gas is to take place. Appropriate valving will include valves provided at A, B, C and D for appropriate control and isolation means. Linked control systems 5 and 6 monitor and maintain the supply to and the conditions in the chambers 3 and 4.

From the process chamber 4 gases pass to an exhaust system 7, which in turn leads to an abatement tool 8 (which is usually needed). A bypass outlet 9 leads from the reaction chamber 3 to the exhaust system, whereby gases can be switched into the process chamber only when required for processing. This also allows the means for ensuring stable gas composition and flow to be maintained prior to switching into the process chamber.

Figure 2 shows features of the reaction chamber 3, and connections thereto, in greater detail. This illustrates inlets 1A and 2A from sources 1 and 2 and another inlet 10 for a third precursor gas (there could be still further inlets if required). A control system 11 (which is a single system but for clarity is shown in two parts 11 and 15) can take measurements from analysis devices 12 and 13 and a pressure-measuring device 14. The control system 15 can take measurements from a temperature measurement device 16 and also provide control for an auto pressure control device 17 and a temperature control device 18. Optionally an electrode 19 can be provided within the reaction chamber 3 with biasing means (not shown) and a temperature control

device (not shown).

Figures 3, 4, 5 and 6 illustrate various ways in which electrodes can be incorporated into the reaction chamber 3 (which in turn may comprise a single chamber or a combination of chambers linked in serial or parallel or any desired combination thereof). In Figure 3 the chamber 3 is shown as being provided with a porous electrode 20. Figure 4 illustrates a two chamber system wherein the preliminary reaction chamber 3A incorporates a normal electrode 21 whilst a downstream chamber 3B incorporates a porous electrode 22. The third gas supply line 10 is connected to this secondary chamber 3B. In Figure 5 three linked reaction chambers 3A, 3B and 3C are shown, each incorporating a standard electrode 23, 24 and 25 with respective gas supply sources 3, 4 and 10. In the modified arrangement shown in Figure 6 the chambers 3A, 3B and 3C are modified to incorporate porous electrodes 23A, 24A and 25A.

The procedures of this invention are particularly useful additionally as a method of local creation of a number of short shelf-life organo-metallic compounds using suitable precursors.

CLAIMS

1. A gas generator system comprising supply sources for halogenated gases, including pure molecular halogens, connected into a gas reaction chamber or chamber system  
5 enabling generation of a predetermined gas for localised use in a subsequent process.
2. A gas generator system as claimed in claim 1, wherein the reaction chamber has a valved outlet for direct supply of the generated gas to a process chamber or a single or  
10 multiple chamber processing tool or tools.
3. A gas generator system as claimed in claim 1 or claim 2, which provides for the localised generation of reactive process gases.
4. A gas generator system as claimed in claim 1 or claim  
15 2, which provides for the local synthesis of reactive precursors for plasma and non-plasma processing which include precursors which would otherwise not readily available due to instability or short-lifetime or difficulty in commercial manufacture or supply.
- 20 5. A gas generator system as claimed in claim 1 or claim 2, which provides for the synthesis of precursors that replace directly or indirectly (by simulating the behaviour) the gases that are either in use or have been used in industry and are subject to restriction in use or  
25 withdrawal by environmental legislation.
6. A method of generating gases for localised use in a subsequent process by utilising the system as defined in



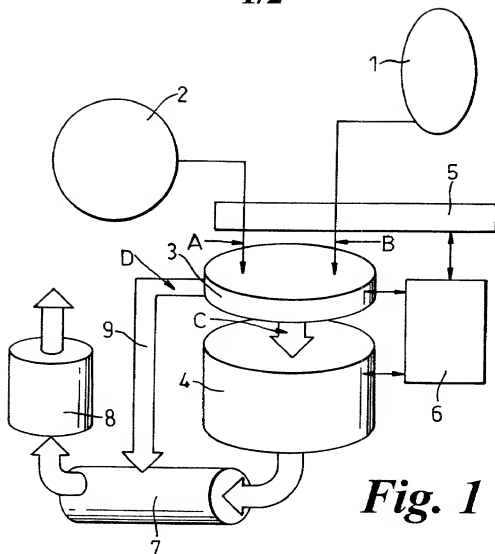
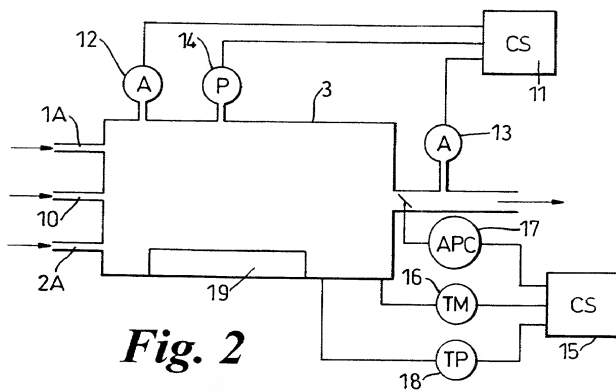
any one of claims 1 to 5 and using the gas so generated directly in the subsequent process.

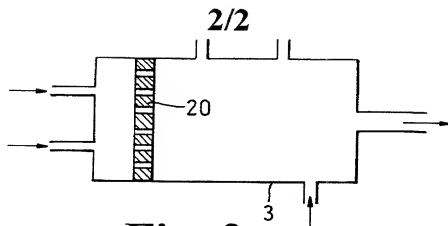
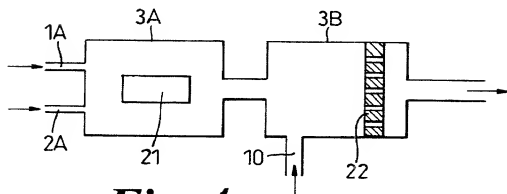
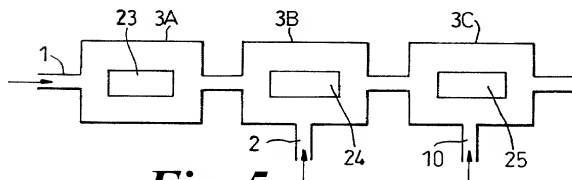
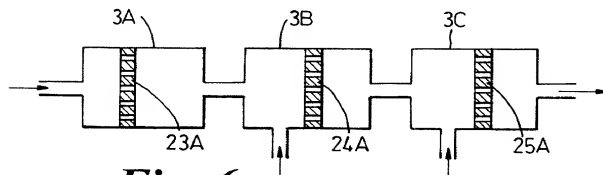
7. A gas generator system according to claim 1 and substantially as herein described.

5 8. A method of generating gases according to claim 6 and substantially as herein described.

9. Any novel combination of a gas generator or a process for generating gases and with reference to the accompanying drawings.

1/2

**Fig. 1****Fig. 2**

**Fig. 3****Fig. 4****Fig. 5****Fig. 6**

## DECLARATION AND POWER OF ATTORNEY FOR U.S. PATENT APPLICATION

(X) Original    ( ) Supplemental    ( ) Substitute    ( ) PCT    ( ) Design

As a below named inventor, I hereby declare that: my residence, post office address and citizenship are as stated below next to my name; that I verily believe that I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural inventors are named below) of the subject matter which is claimed and for which a patent is sought on the invention entitled:

### TITLE: GAS GENERATION SYSTEM

of which is described and claimed in:

( ) the attached specification, *or*

( ) the specification in the application Serial No. \_\_\_\_\_ filed \_\_\_\_\_,  
and with amendments through \_\_\_\_\_ (if applicable), *or*

(X) the specification in International Application No. PCT/ GB00/00793 , filed 6 MARCH 2000 ,  
and as amended on \_\_\_\_\_ (if applicable).

I hereby state that I have reviewed and understand the content of the above-identified specification, including the claims, as amended by any amendment(s) referred to above.

I acknowledge my duty to disclose information of which I am aware which is material to the examination of this application in accordance with Title 37, Code of Federal Regulations, §1.56(a).

I hereby claim foreign priority benefits under Title 35, United States Code, §119 (and §172 if this application is for a Design) of any foreign application(s) for patent or inventor's certificate listed below and have also identified below any foreign application for patent or inventor's certificate having a filing date before that of the application on which priority is claimed:

COUNTRY	APPLICATION NO.	DATE OF FILING	PRIORITY CLAIMED
GREAT BRITAIN	9904925.6	4 MARCH 1999	YES
GREAT BRITAIN	9909853.5	29 APRIL 1999	YES

I hereby claim the benefit under Title 35, United States Code, §120 of any United States application(s) listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States application in the manner provided by the first paragraph of Title 35, United States Code, §112, I acknowledge the duty to disclose material information as defined in Title 37, Code of Federal Regulations, §1.56(a) which occurred between the filing date of the prior application and the national or PCT international filing date of this application:

APPLICATION SERIAL NO.	U.S. FILING DATE	STATUS: PATENTED, PENDING, ABANDONED

And I hereby appoint Raymond C. Jones, Reg. No. 34,631 and Adam C. Volentine, Reg. No. 33,289, of the firm of JONES VOLENTINE, L.L.C., jointly and severally, attorneys to prosecute this application and to transact all business in the U.S. Patent and Trademark Office connected therewith.

I hereby authorize the U.S. attorneys named herein to accept and follow instructions from WYNNE-JONES LAINE & JAMES as to any action to be taken in the U.S. Patent and Trademark Office regarding this application without direct communication between the U.S. attorneys and myself. In the event of a change in the persons from whom instructions may be taken, the U.S. attorneys named herein will be so notified by me.

Kindly direct all correspondence to:

**JONES VOLENTINE, L.L.C.**  
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Suite 150  
Reston, Virginia 20191  
 Telephone (703) 715-0870

Full Name of 1st Inventor	FAMILY NAME <u>BHARDWAJ</u>	FIRST GIVEN NAME <u>Jyoti</u>	SECOND GIVEN NAME <u>Kiron</u>
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Full Name of 2nd Inventor	FAMILY NAME	FIRST GIVEN NAME	SECOND GIVEN NAME
Residence & Citizenship	CITY	STATE OR COUNTRY	COUNTRY OF CITIZENSHIP
Post Office Address	ADDRESS	CITY	STATE OR COUNTRY ZIP CODE

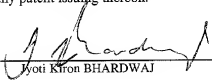
Full Name of 3rd Inventor	FAMILY NAME	FIRST GIVEN NAME	SECOND GIVEN NAME
Residence & Citizenship	CITY	STATE OR COUNTRY	COUNTRY OF CITIZENSHIP
Post Office Address	ADDRESS	CITY	STATE OR COUNTRY ZIP CODE

Full Name of 4th Inventor	FAMILY NAME	FIRST GIVEN NAME	SECOND GIVEN NAME
Residence & Citizenship	CITY	STATE OR COUNTRY	COUNTRY OF CITIZENSHIP
Post Office Address	ADDRESS	CITY	STATE OR COUNTRY ZIP CODE

Full Name of 5th Inventor	FAMILY NAME	FIRST GIVEN NAME	SECOND GIVEN NAME
Residence & Citizenship	CITY	STATE OR COUNTRY	COUNTRY OF CITIZENSHIP
Post Office Address	ADDRESS	CITY	STATE OR COUNTRY ZIP CODE

I further declare that all statements made herein of my own knowledge are true, and that all statements on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code, and that such willful false statements may jeopardize the validity of the application or any patent issuing thereon.

1st Inventor  Date 23<sup>rd</sup> October 2000.  
Jyoti Kiron BHARDWAJ  
 2nd Inventor \_\_\_\_\_ Date \_\_\_\_\_  
 3rd Inventor \_\_\_\_\_ Date \_\_\_\_\_  
 4th Inventor \_\_\_\_\_ Date \_\_\_\_\_  
 5th Inventor \_\_\_\_\_ Date \_\_\_\_\_

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Applicant Reference No.: STS.32 Atty Docket No.: WLJ.059